Treatment of Emerging Chemical and Microbial Contaminants in Water using Advanced Reflective UV Technology

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### Emerging Contaminants (ECs) Xenobiotics

#### **Personal Care Products**



#### **Industrial By-Products and Effluents**











**Pharmaceutical Products** 



Paints/Dyes, Deodorants, and Pesticides

# Nitrosodimethylamine (NDMA)

- Belongs to a highly potent family of carcinogens nitrosamines
- US DHHS reports NDMA to form tumors in kidneys, liver, blood vessels and trachea in experimental animals (IARC,1999)
- Potency to cause cancer at low concentrations of 0.7 ng/L makes it more potent than trihalomethanes (Mitch, Gerecke et al., 2003)
- Is listed in CCL4 (2016)
- No regulatory limit yet



### **Conventional Water Treatment Techniques - NDMA**

- UV irradiation is the most applied treatment technology
- Low radical conversion rate and low hydroxyl stability of H<sub>2</sub>O<sub>2</sub> makes it selectively effective in certain water reuse trains (Fujioka et al., 2012; Milkos et al., 2018)
- Chlorine should not be used because of the high chances of NDMA reoccurrence
- Sulfate radicals (peroxydisulphate PDS) have been found to be effective in the treatment of NDMA but they are known have higher sensitivity to changes in the water matrix (Feng et al., 2017)
- No studies with AOP under dynamic/flow conditions.



# 1, 4 – Dioxane (1, 4 – D)

- Synthetic organic cyclic diether
- Highly miscible in water and thus non-volatile when in water
- Not easily bio-degraded
- Classified as a carcinogen to humans by all routes of exposure (U.S. EPA, 2013a)
- Is listed in CCL4 (2016)
- No regulatory limit yet





# Conventional Treatment Techniques: 1, 4 – D

- Low Henry's constant and BP of 101°C makes physical treatment systems like thermal destruction, air stripping and distillation inefficient (Vescovi et al., 2010; Mohr, 2010)
- With UV absorption peak of 165 191nm, direct photolysis is inefficient (Pickett et al., 1951)
- UV/H<sub>2</sub>O<sub>2</sub>: 99% remediation in 10h reaction time.
- Doses of 20 mg/L for UV/H<sub>2</sub>O<sub>2</sub> and 6, 1.5 mg/L respectively for O<sub>3</sub>/H<sub>2</sub>O<sub>2</sub> to obtain 80% remediation (Ikehata et al., 2016)
- No dynamic/flow studies reported using AOP







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# PAA verses H<sub>2</sub>O<sub>2</sub>

- $H_2O_2$  is commercially subjected to acid catalysis to form PAA.
- Ionization potential
  - H<sub>2</sub>O<sub>2</sub> = 1.78eV
  - PAA = 1.96eV
- PAA has higher hydroxyl stability than  $H_2O_2$ .
- Radicals formed upon UV irradiation
  - H<sub>2</sub>O<sub>2</sub>: hydroxyl radicals
  - PAA: acetyloxy , methyl, weak peroxyl, acetylperoxyl and hydroxyl radicals



NDMA: A Threat and an Emerging Contaminant – Control Strategies using Peracetic Acid with Low Pressure UV and Reflective Technology



# Nitrosodimethylamine (NDMA)

- Semi volatile
- Slightly yellow, water soluble, odorless
- UV absorption peak is at 222 nm
- Hepatotoxin and carcinogen at 0.7 ng/L
- Remediation by conventional treatment train or GACs is inefficient





### Experimental Setup



The UV devices were generously provided by NeoTech Aqua Solutions, San Diego, CA

Influent

PAA

Effluent samples were analyzed by Scottsdale Water Campus in accordance with EPA 521 method to analyze nitrosamines.

Effluent





# **Experimental Variables**

PAA (mg/L)	0.5	1	1.5	2	3	4
Flowrate (GPM)	1	1.2	5	10	20	0
UV Wavelength (nm)		254			220	
Influent NDMA (ng/L)			<b>150 t</b>	o 650		



### AOP of NDMA using UV 220 nm



 NDMA can be efficiently remediated using low doses of UV 220 nm



# Effect of PAA concentration on AOP of NDMA using UV at 220 nm



- PAA had a significant effect on NDMA remediation
- PAA > 1.5 2 mg/L had a negative impact on NDMA remediation



### NDMA remediation using UV 254 nm



- NDMA remediation is directly dependent to UV dose
- The highest remediation was 94.2% using >1000 mJ/cm<sup>2</sup> UV



### Effect of PAA dose on AOP of NDMA using UV at 254 nm



• NDMA remediation improved with increase in PAA dose



 UV dose > 300 mJ/cm<sup>2</sup>, PAA at 1.5 mg/L resulted in decrease of percent remediation



### Effect of PAA dose on AOP of NDMA using UV at 254 nm



- PAA at 1.5 mg/L consistently showed decreased NDMA remediation for UV dose >300 mJ/cm<sup>2</sup>
- PAA dose had minimal impact on NDMA remediation for UV >1000 mJ/cm<sup>2</sup>

### Conclusion - Considerations

- 220 vs 254 nm
- PAA vs H<sub>2</sub>O<sub>2</sub>
- Low vs High UV dose
- Optimum PAA concentration 0.5 mg/L (254 nm), 1.5 mg/L (220 nm)
- Optimum UV dose 245 mJ/cm<sup>2</sup> (254 nm lamp), 5.5 mJ/cm<sup>2</sup> (220 nm lamp)



# AOP Involving Reflective UV Technology and Peracetic Acid : A Promising Alternate Technology in the Treatment of 1, 4 – Dioxane in Water



# 1, 4 – Dioxane (1, 4 – D)

- Cyclic diether, completely miscible in water and organic solvents.
  UV absorption peak of 165 to 191 nm
- IARC has classified this as a Group 2B agent (potential carcinogen pending data on humans) (IARC, 1999)
- The U.S. EPA has also recognized and classified this compound as a likely carcinogen to humans (U.S. EPA, 2013b)
- The USEPA has a Health Advisory Level (HAL) of 0.35 μg/L in drinking water (U.S. EPA, 2013b)



### **Experimental Setup**





The UV devices were generously provided by NeoTech Aqua Solutions, San Diego, CA Effluent samples were analyzed by Scottsdale Water Campus in accordance with EPA 522 method to analyze 1, 4 – D in drinking water

Effluent



Variables	Parameters Tested							
PAA (mg/L)	1	2		3		4	5	7
Flowrate (GPM)	1	2.5	5		4	1	0	20
UV wavelength (nm)	254							
Influent 1,4 - D (mg/L)	5 to 45							



### Remediation of 1, 4 – D using UV 254 nm



- UV doses >1000 mJ/cm<sup>2</sup> resulted in high percent remediation
- 1, 4 D remediation decreased at reduced UV doses



### Remediation of 1,4 – D using PAA



PAA showed potential as a catalyst in AOP involving UV 254 nm



### AOP of 1,4 – Dioxane using UV/PAA



AOP involving UV at 790 and 460 mJ/cm<sup>2</sup> with PAA at 5 mg/L yielded in optimized remediation



# Conclusion

- The dependency of the remediation of 1,4 dioxane on PAA dose and UV dose was established.
- UV 254nm yielded 92% remediation at 1000mJ/cm<sup>2</sup>
- PAA yielded 14.5% remediation at 5mg/L
- UV/PAA yielded the highest remediation
  - 1. 99.6% at 1,000 mJ/cm<sup>2</sup> and 5 mg/L PAA
  - 2. 77% at 420 mJ/cm<sup>2</sup> and 5 mg/L PAA
- At PAA concentrations higher than 5mg/L, high degree of UV scavenging was observed.

Note: 1,000 mJ/cm<sup>2</sup> was achieved at 1 GPM flowrate 420 mJ/cm<sup>2</sup> was achieved at 4 GPM flowrate



Microbial contaminants in water : Inactivation of *E. coli, Legionella, Mycobacterium* and Fungal spores using UV with reflective technology and AOP involving UV/PAA



### Low-Pressure UV 254 nm with Reflective Technology

<u>E. coli</u>	<u>Legionella</u>	<u>Aspergillus niger</u>	<u>Mycobacterium</u>
	pnemophila	Spores	<u>avium</u>
> 7 log inactivation with 115 mJ/cm <sup>2</sup> UV	> 5 log inactivation with 125 mJ/cm <sup>2</sup> UV	> 2 log inactivation AOP: 530 mJ/cm <sup>2</sup> UV plus 2mg/L PAA	Currently in Progress







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